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32. State Boyle's Law and Charle's Law. Derive the relation $Pv=nRT$. An evacuated glass vessel weighs 50 gm when empty, 148 gm when filled with a liquid of density 0.98 gm/ml, and 50.5 gm when filled with an ideal gas at 760 mmHg at 27 °C. Determine the molecular mass of the gas.

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MEMORANDUM: Maskew Miller Longman, Focus on Physical Sciences Grade 11 – Chemistry: Paper 2 Page 4 of 10 Section B Question 4 4.1 What is the relationship between volume and temperature of a given mass of gas that is kept at constant pressure? [11.1.1] (2)

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