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Solution This problem is solved using two steps: (1) calculate the total number of lattice sites in silver, N , using Equation 4.2, and (2) take the ratio of the equilibrium number of vacancies given in the problem statement (N

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If the ionic radius of the cation is 0.080 nm, what is the anion radius? Solution

To begin, let us rewrite Equation 2.15 to read as follows: $r_0 = r_C + r_A$ in which r_C and r_A represent, respectively, the radii of the cation and anion. Thus, this problem calls for us to determine the value of r_A .

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